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1 Introduction 1.1 A gas at 20 ° C

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may be rarefied if it contains less than 10^{12} molecules per mm^3 . If Avogadro's number is 6.023×10^{23} molecules per mole, what air pressure does this represent?

Solution: The mass of one molecule of air may be computed as Molecular weight 28.97 mol^{-1}

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Solution 1.1. To get started, first list or determine the volumes

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involved: d = volume of water
dumped = 100 cm^3 , c = volume
of a sip = 5 cm^3 , and $V_2 =$
volume of water in the oceans = $\frac{4}{3} \pi R^2 D$, where, R is the radius of
the earth, D is the mean depth of
the oceans, and f is the oceans'
coverage fraction. Here we've

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ignored the ocean volume occupied
by salt and have assumed that the
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Solution 1.1. To get started, first list or determine the volumes

involved: d = volume of water dumped = 100 cm^3 , c = volume of a sip = 5 cm^3 , and V = volume of water in the oceans = $\approx 4 \times 10^{21} \text{ m}^3$

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D , where, R is the radius of the earth, D is the mean depth of the oceans, and is the oceans' coverage fraction. Here we've ignored the ocean volume occupied by salt and have assumed that the ...

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